

Material Safety Data Sheet

1 Identification of substance:

• Product name: Stainless Steel Weld Neutraliser

• Stock number: 129934

• Manufacturer/Supplier:

PPS Industries Limited 86 Hugo Johnston Drive,

Penrose,

Auckland, New Zealand

P.O.Box 12-823, Penrose, Auckland 1061

Phone: 64 9 579-1001 Facsimile: 64 9 579-9098

Emergency Phone: 0800 657 894 Monday to Friday 8am-4pm

Web Site: www.ppsindustries.co.nz

Emergency contact detail:

For emergency only. During normal hours call PPS Industries office.

OrganizationLocationPhoneNationalPOSION CENTERNew Zealand0800 764-766Chemcall24/7 Emergency Response ServiceNew Zealand0800 243-622

• 2 Hazards identification

Classified as hazardous according to the criteria in the Hazardous Substances (Minimum Degrees of Hazard) Regulations 2001. Classified as a Dangerous Goods according to NZS 5433.

- Hazard description: Class 8, Sodium Hydroxide Solution, Packing Group II, UN 1824
- HSNO Class:

Class 6 Toxicity

6.1D(oral) Acutely Toxic.

6.1E(dermal) Acutely Toxic.

Class 8 Corrosive

8.1A Corrosive to metal.

8.2B Corrosive to dermal tissue.

8.3A Corrosive to ocular tissue.

Class 9 Ecotoxicity

- 9.1D(fish) Slightly harmful in the aquatic environment or are otherwise designed for biocidal action.
- 9.1D(crustacean) Slightly harmful in the aquatic environment or are otherwise designed for biocidal action.
- EPA Approved Number: HSR001576
- GHS Classification:

Hazard Pictogram:





Signal word: Danger

Hazard class:

Hazardous to the Aquatic environment Long term hazard Category 4
Acute Toxicity: Oral Category 4
Acute Toxicity: Dermal Category 5
Skin corrosion/irritation Category 1B
serious eye damage/eye irritation Category 1
Corrosive to metals category 1

Hazard statement(s):

H413 May cause long lasting harmful effects to aquatic life

H302 Harmful if swallowed

H313 May be Harmful in contact with skin

H314 Causes skin burns and eye damage

H318 Causes serious eye damage

H290 May be corrosive to metals

Precautionary Statement(s) Prevention:

P234 Keep only in original container

 $\ensuremath{\mathsf{P}} 101\ \ensuremath{\mathsf{If}}$ medical advice is needed, have product container or label at hand.

P102 Keep out of reach of children

P103 read labels before use.

P264 Wash the hands and other exposed parts of the body, thoroughly after Handling.

P270 Do not eats, drink or smoke when using this product.

P260 Do not breathes dust/fume/gas/mist/vapours/spray.

P264 Wash the affected body parts thoroughly after handling.

P280 wear protective gloves/protective clothing/eye protection/face protection.

P273 Avoid release to the environment

Precautionary Statement(S) Response:

P390 Collect spillage

 ${\tt P301+P310}$ If swallowed: Immediately call poison centre or doctor. Rinse Mouth With water.

P312 Call a poison centre or doctor if you feel unwell

P301+P330+P331 If swallowed: Rinse mouth. Do not induce vomiting.

P303+P361+P353 If on skin (or hair): Take off immediately all

contaminated clothing. Rinse skin with water (or shower)

P363+P304+P340 Wash contaminated clothing before reuse. If inhaled:

Remove victim to fresh air and keep at rest in a position comfortable for breathing.

P310 Immediately call a poison centre or a doctor.

P321 For specific treatment read the Label.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several



minutes. Remove contact Lenses if present and easy to do. Continue rinsing.

P310 Immediately call a POISON CENTER or doctor/physician.

Precautionary Statement(S)Storage:

P405 Store locked up.

P404 Store in a closed container

Precautionary Statement(S) Disposal:

P501 Dispose of contents/container to an appropriate treatment and disposal Facility in accordance with applicable laws and regulations.

3 Composition/Data on components:

• Chemical characterization:

Description:	(CAS#)	Concentration	Hazardous
Sodium Hydroxide	1310-73-2	<30 % w/w	Yes
Water	7732-18-5	balance	No

4 First aid measures

• After inhalation

Supply fresh air. If required, provide artificial respiration. Keep patient warm.

Seek immediate medical advice.

• After skin contact

Immediately wash with water and soap and rinse thoroughly. Seek immediate medical advice.

After eye contact

Rinse opened eye for several minutes under running water. Then consult a doctor.

After swallowing

DO NOT INDUCE VOMITING! Give large quantities of water or milk if available. Never give anything by mouth to an unconscious person. Get medical attention immediately.

Information for doctor

Perform endoscopy in all cases of suspected sodium hydroxide ingestion. In cases of severe esophageal corrosion, the use of therapeutic doses of steroids should be considered. General supportive measures with continual monitoring of gas exchange, acid-base balance, electrolytes, and fluid intake are also required.

5 Fire fighting measures

Fire

Not considered to be a fire hazard. Hot or molten material can react violently with water.

Can react with certain metals, such as aluminium, to generate flammable hydrogen gas.

Explosion:

Not considered to be an explosion hazard.

Fire Extinguishing Media:

Use any means suitable for extinguishing surrounding fire. Adding water to caustic solution generates large amounts of heat.



Special Information:

In the event of a fire, wear full protective clothing and approved self-contained breathing apparatus with full facemask operated in the pressure demand or other positive pressure mode.

6 Accidental release measures

Person-related safety precautions:

Wear protective equipment. Keep unprotected persons away. Ensure adequate ventilation

Measures for environmental protection:

Do not flush caustic residues to the sewer. Residues from spills can be diluted with water, neutralized with dilute acid such as acetic, hydrochloric or sulphuric.

Measures for cleaning/collecting:

Absorb neutralized caustic residue on clay, vermiculite or other inert substance and package in a suitable container for disposal.

Additional information:

See Section 7 for information on safe handling See Section 8 for information on personal protection equipment. See Section 13 for disposal information.

7 Handling and storage

Handling

Information for safe handling:

Keep container tightly sealed. Store in cool, dry place in tightly closed containers. Ensure good ventilation at the workplace. Always add the caustic to water while stirring; never the reverse.

Storage

Requirements to be met by storerooms and receptacles:

Store in a cool location.

Information about storage in one common storage facility:

Do not store with aluminium or magnesium. Do not mix with acids or organic materials. Store away from water/moisture.

8 Exposure controls and personal protection

- A system of local and/or general exhaust is recommended to keep employee exposures below the Airborne Exposure Limits. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area.
- Personal protective equipment
- Breathing equipment:



Use suitable respirator when high concentrations are present.

• Protection of hands: Impervious gloves

• Eye protection: Safety glasses

Body protection: Protective work clothing.

9 Physical and chemical properties:

Appearance: White, deliquescent pellets or

flakes.

Odour: Odourless.

Solubility: 100 g/100 g of water.

Specific Gravity: 1.3

pH: 13 - 14 (0.5% soln.)
Boiling Point: No information found.
Melting Point: No information found.

10 Stability and reactivity

Stability:

Stable under ordinary conditions of use and storage. Very hygroscopic. Can slowly pick up moisture from air and react with carbon dioxide from air to form sodium carbonate.

Hazardous Decomposition Products:

Sodium oxide. Decomposition by reaction with certain metals releases flammable and explosive hydrogen gas.

Hazardous Polymerisation:

Will not occur.

Incompatibilities:

Sodium hydroxide in contact with acids and organic halogen compounds, especially trichloroethylene, may causes violent reactions. Contact with nitromethane and other similar nitro compounds causes formation of shock-sensitive salts. Contact with metals such as aluminium, magnesium, tin, and zinc cause formation of flammable hydrogen gas. Sodium hydroxide, even in fairly dilute solution, reacts readily with various sugars to produce carbon monoxide. Precautions should be taken including monitoring the tank atmosphere for carbon monoxide to ensure safety of personnel before vessel entry.

Conditions to Avoid:

Moisture, dusting and incompatibles.

• 11 Toxicological information

Irritation data: skin, rabbit: 500 mg/24H severe; eye rabbit: 50 ug/24H severe; investigated as a mutagen.

-----\Cancer Lists\-----

---NTP Carcinogen---

12 Ecological information:

Environmental Fate:

No information found.

Environmental Toxicity:



No information found.

13 Disposal considerations

- Product:
- Recommendation

Whatever cannot be saved for recovery or recycling should be handled as hazardous waste and sent to a approved waste facility. Consult state, local or national regulations to ensure proper disposal.

- Uncleaned packagings:
- Recommendation:

Disposal must be made according to official regulations.

14 Transport information

DOT regulations:

• Hazard class:

Identification number: UN1824Packing group: II

• Proper shipping name (technical name):

SODIUM HYDROXIDE SOLUTION

15 Regulations

• HSNO Class:

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Certified Handler : Not ApplicableTracking : Not Applicable

16 Other information:

Employers should use this information only as a supplement to other information gathered by them, and should make independent judgement of suitability of this information to ensure proper use and protect the health and safety of employees. This information is furnished without warranty, and any use of the product not in conformance with this Material Safety Data Sheet, or in combination with any other product or process, is the responsibility of the user.



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